



A Straightforward Synthesis of (2*S*,3*R*)-3-Hydroxyproline and *trans*-(2*R*,3*S*)-2-Hydroxymethyl-3-hydroxypyrrolidine

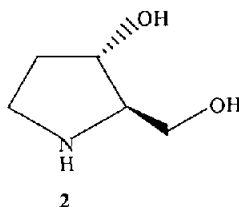
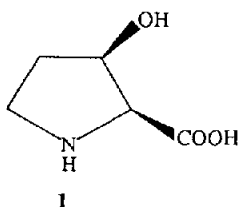
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Abstract: A stereocontrolled synthesis of (2*S*,3*R*)-3-hydroxyproline **1**, and *trans*-(2*R*,3*S*)-2-hydroxymethyl-3-hydroxypyrrolidine **2** has been achieved in 21% and 38% yield *via* the homochiral 4,5-disubstituted oxazolidin-2-one **3**. The *trans* relationship in **2** has been introduced by a modified Mitsunobu reaction.

(2*S*,3*R*)-3-Hydroxyproline **1**, formally *cis*-3-hydroxy-L-proline is a known, although rare, β-hydroxy-α-amino acid, which has been found as a component of the antibiotic teleomycin.¹ It has been isolated as minor component in collagenous proteins from carcinoma cell cultures,² and it is used in pharmaceutical preparations for antitumor therapy³ or for treatment of disorders of collagen metabolism.⁴ It can be easily converted into the Geissman-Waiss lactone,⁵ which is a key intermediate for the preparation of a variety of pyrrolizidine alkaloids, including (+) retronecine,^{5,6} (-) platynecine,⁶ and (+) croalbinecine.⁶ The corresponding protected *cis*-3-hydroxy-L-prolinol has been used by Joullie *et al.* as an intermediate in the synthesis of non-proteinogenic amino acids as detoxinine⁷ and in the synthesis of munnularine F.⁸



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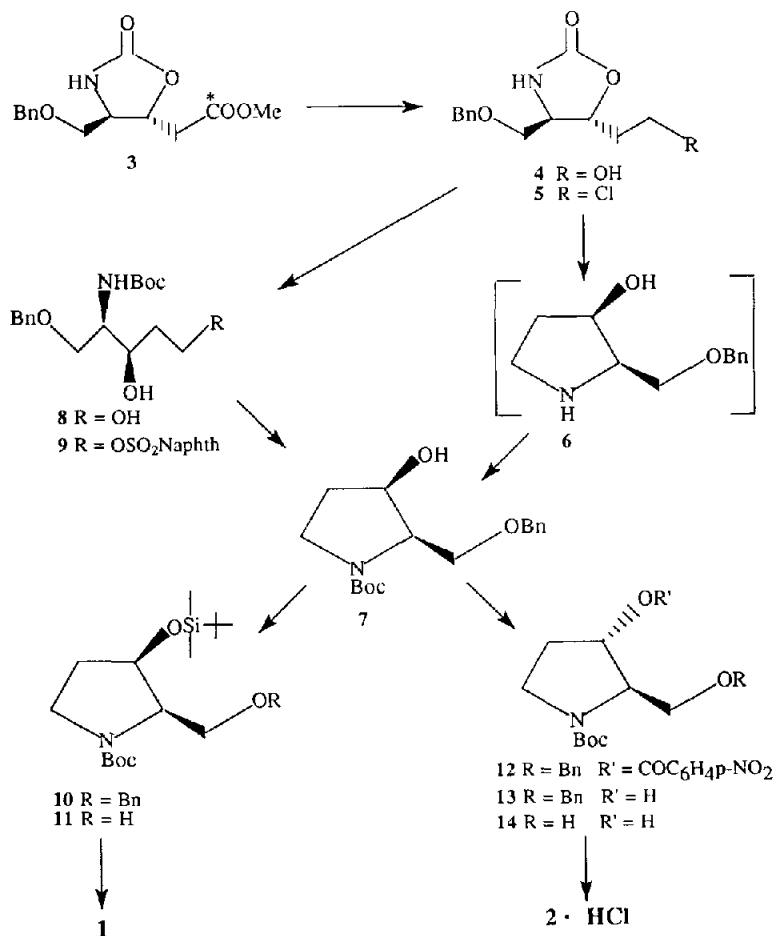
The isomer (2*R*,3*S*)-3-hydroxyproline and the corresponding (2*R*,3*R*)-3-hydroxyprolinol are structural units in indolizidine alkaloids as slaframine⁹ and castanospermine¹⁰ and in pyrrolizidine alkaloids, as australine¹¹ and alexine.¹²

Moreover, the structurally related (2*R*,3*S*)-2-hydroxymethyl-3-hydroxypyrrolidine, or *trans*-3-hydroxy-L-prolinol **2**, occurs in the seeds of the legume *Castanospermum australe*.¹³

Several syntheses of *cis*-3-hydroxyproline¹⁴ and *trans*-3-hydroxy-L-prolinol¹⁵ have been developed. This paper deals with the preparation of the title compounds **1** and **2** through the key intermediate *cis*-3-hydroxypyrrolidine **7**, where the pyrrolidine nucleus is formed by ring closure between the indicated (*) carbon atom and the nitrogen atom in the 4,5-disubstituted oxazolidin-2-one **3**.

We have recently reported¹⁶ a simple and convenient route to **3** *via* an highly stereoselective cyclocarbamation, mediated by iodine, of the methyl (2*Z*,4*R*)-5-benzyloxy-[(benzyloxycarbonyl)amino]-2-pentenoate, readily available from L-serine,¹⁷ followed by tributyltin hydride reduction.

According to the synthetic pathway shown in the Scheme the ester group in **3** was reduced by the use of NaBH₄ in THF/MeOH¹⁸ in 88% yield (this is more convenient than the already reported reduction with LiAlH₄^{16a}).



The alcohol **4** was initially converted into the chloride **5**, m.p. 51-53, $[\alpha]_D^{25} = +67.5$ ($c = 0.25$, CHCl_3), in 93% yield, following standard methodology¹⁹ ($\text{Ph}_3\text{P}/\text{CCl}_4$), and then treated with a solution of NaOH in $\text{MeOH}/\text{H}_2\text{O}$ (80°C), to give the pyrrolidine **6**, through cleavage of the cyclic urethane and displacement of chlorine by nitrogen. The intermediate **6** was directly converted into the *N*-Boc derivative **7**, $[\alpha]_D^{25} = -20.0$ ($c = 1.88$, MeOH) (65% yield based on **5**), by reaction with Boc_2O in the presence of Et_3N , giving a 61% yield from **4**.

In a less convenient way, the oxazolidin-2-one **4** was initially treated with $\text{Ba}(\text{OH})_2$ in MeOH at reflux to cleave the cyclic carbamate, and then converted into the *N*-Boc amino alcohol **8** in 68%. Regioselective activation of the primary alcohol as β -naphthalensulphonate **9** and ring closure (57% based on **8**) with NaH , gave the desired pyrrolidine **7**, but only in a 40% yield from **4**.

Compound **7** was readily converted into the corresponding *tert*-butyldimethylsilyl ether **10**, followed by removal of the benzyl protecting group by catalytic hydrogenolysis over 10% Pd/C to give the known 2-hydroxymethylpyrrolidine **11**,²⁰ $[\alpha]_D^{25} = -33.1$ ($c = 1.2$, CHCl_3) (Lit²⁰: $[\alpha]_D^{25} = -32.5$ ($c = 1.2$, CHCl_3)), in 63% yield from **7**. Finally the (2*S*,3*R*)-3-hydroxyproline **1** was obtained by a two step oxidation (TEMPO , NaClO/NaBr ²¹ and 1 M $\text{KMnO}_4/5\% \text{NaH}_2\text{PO}_4$ ²²) to give the corresponding acid, which upon treatment with HCl/MeOH afforded the target β -hydroxy- α -amino acid **1**, m.p. 222-228°C, $[\alpha]_D^{25} = -100.4$ ($c = 1.0$, H_2O) (Lit^{14e}: $[\alpha]_D^{25} = -101$ ($c = 1.0$, H_2O)), in 21% yield from **3**.

The (2*R*,3*S*)-2-hydroxymethyl-3-hydroxypyrrolidine **2** was obtained from **7** through the following steps. The inversion of the alcohol configuration was achieved by the Mitsunobu reaction,²³ with the modifications (*p*- $\text{NO}_2\text{C}_6\text{H}_4\text{COOH}$, diethylazodicarboxylate, Ph_3P , benzene, 25°C) reported by Martin *et al.*²⁴ for secondary alcohols, because under standard Mitsunobu conditions only starting material has been obtained. The *p*-nitrobenzoate **12** was easily obtained in 90% yield with inversion of configuration, and it was transformed into the *trans*-3-hydroxypyrrolidine **13**, m.p. 67-68°C, $[\alpha]_D^{25} = -33.4$ ($c = 0.18$, MeOH), in 90% yield under alkaline conditions.

The *cis*- and *trans*-relationships between the two substituents in hydroxyprolines **7** and **13** were substantiated by the values of the coupling constants of H-2 and H-3 protons in ^1H NMR spectra ($J_{2,3} = 6.5$ Hz and 1.5 Hz, respectively).^{14h} These data were confirmed by difference NOE experiments, which showed in particular the proximity of H-2 α and H-3 α in compound **7** and of H-3 β and CH_2O in compound **13**.

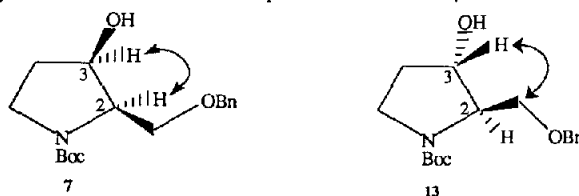


Fig.: Mutual NOE effects in compounds **7** and **13**

The ^1H - and ^{13}C NMR spectra of *N*-Boc protected pyrrolidines were not interpretable at room temperature for the presence of a dynamic equilibrium between two rotamers due to the restricted rotation of the nitrogen-carbon bond of the amide group. To overcome this phenomenon NMR spectra were run at 50°C. The chemical shift assignments in ^1H NMR spectra were unambiguously confirmed by decoupling experiments and/or HECTOR measurements. Conversely, in ^{13}C NMR spectra two signals of comparable intensity were still present for most of the carbons and both are reported in the Experimental section. No attempt was made to assign the resonances of each rotamer.

The intermediate **13** was quantitatively converted, by catalytic hydrogenolysis over 10% Pd/C , into the

diol **14**, which was then treated with 3 M HCl/EtOAc²⁵ to give **2** as hydrochloride, $[\alpha]_D = +43.1$ ($c = 0.5$, H₂O) (Lit.¹³: $[\alpha]_D = +46.5$ ($c = 1.0$, H₂O)) in 38% yield from **3**.

In summary, we have reported on the easy preparation of (2*S*,3*R*)-3-hydroxyproline and (2*R*,3*S*)-2-hydroxymethyl-3-hydroxypyrrolidine. The present method proves to be a suitable alternative to the known procedures; moreover since the isomer **3** [(4*S*,5*S*)] is readily obtained in optically pure form,^{16b} our methodology could be followed to prepare the (2*R*,3*S*)-3-hydroxyproline. Further applications of the above protected pyrrolidines in the total synthesis of biologically relevant alkaloids are currently under investigation.

Experimental

Melting points (uncorrected): open capillaries, Büchi apparatus. – IR spectra (films or in KBr for solids): Nicolet 5DX FT-IR. – ¹H- and ¹³C-NMR (300 MHz and 75 MHz, TMS as internal standard in CDCl₃ solutions at 50°C, unless reported otherwise): Varian Gemini 300 spectrometer. – Optical rotations: Perkin-Elmer 243 (measured at 25°C). – Flash chromatography: Merck Kieselgel (particle size 230–400 mesh). All solvents were dried²⁶ prior to use.

(4*R*,5*R*)-4-(Benzyloxy)methyl-5-(2-hydroxyethyl)-oxazolidin-2-one (4): To a stirred and cooled (–10°C,) solution of the ester **3** (1.925 g, 6.9 mmol) in THF (24 ml) was added NaBH₄ (1.03 g, 27.2 mmol) and the mixture was left under stirring for 30 min at the same temperature. After this time MeOH (10 ml) was added dropwise during 30 min and the reaction was stirred at room temp. overnight. H₂O (1 ml) was added carefully and the reaction was stirred for further 30 min. Most of the organic solvent was removed under reduced pressure and brine (10 ml) was added. The mixture was extracted with EtOAc (3 × 100 ml), the combined organic extracts were washed with brine (100 ml) and dried with Na₂SO₄. Evaporation of the solvent under reduced pressure gave a residue which was purified by flash chromatography (CHCl₃/MeOH, 9:1) to give **4** (1.524 g, 88% yield): m.p. 72–73°C, $[\alpha]_D = +23.1$ ($c = 1.8$, CHCl₃); (Lit.^{16a}: m.p. 71–72°C, $[\alpha]_D = +22.8$ ($c = 1.8$, CHCl₃)). ¹H and ¹³C NMR as reported.^{16a}

(4*R*,5*R*)-4-(Benzyloxy)methyl-5-(2-chloroethyl)-oxazolidin-2-one (5): To a solution of the alcohol **4** (0.904 g, 3.6 mmol) in 35 ml of dry CCl₄/CH₂Cl₂ were added successively finely powdered K₂CO₃ (1.0 g, 7.2 mmol) and Ph₃P (2.36 g, 9.0 mmol) and the solution was stirred overnight at 50°C. The solvent was evaporated and the residue was purified by flash chromatography using *n*-hexane/EtOAc (3:7) as eluent to afford the desired alkyl chloride **5** (0.891 g, 93% yield) as a white solid: m.p. 51–53 °C, $[\alpha]_D = +67.5$ ($c = 0.25$, CHCl₃). – IR (KBr): $\nu = 3279$ cm^{–1}, 2846, 1721, 1386, 1242, 1133, 1040, 737. – ¹H NMR: $\delta = 7.40$ – 7.25 (m, 5H, Ph), 5.88 (brs, 1 H, NH), 4.55 (s, 2H, OCH₂Ph), 4.52 (dt, $J = 2 \times 9.0$, 4.5 Hz, 1 H, CHO), 3.69, 3.65 (m, 1 H each, CH₂Cl), 3.67 (dt, $J = 9.0$, 2×6.0 Hz, 1 H, CHN), 3.49 (d, $J = 6.0$ Hz, 2 H, CH₂O), 2.24 (ddt, $J = 15.0$, 9.0, 2×5.0 Hz, 1 H, CH_AH_B), 2.04 (dddd, $J = 15.0$, 8.0, 6.5, 4.5 Hz, 1 H, CH_AH_B). – ¹³C NMR: $\delta = 158.38$ (s, CO), 137.11, 128.57, 128.08, 127.77 (s, 2 × d, d, 2 × d, Ph), 76.14 (d, CHO), 73.59 (t, OCH₂Ph), 71.33 (t, CH₂O), 56.98 (d, CHN), 39.97 (t, CH₂Cl), 37.70 (t, CH₂). Anal. Calcd. for C₁₃H₁₆ClNO₃ (269.7): C, 57.89; H, 5.98; N 5.19. Found: C, 58.21; H, 5.82; N, 5.20.

(2*R*,3*R*)-1-(tert-Butoxycarbonyl)-2-(benzyloxymethyl)-3-hydroxy-pyrrolidine (7): To a solution of chloride **5** (1.076 g, 4.0 mmol) in 10 ml of a 2:1 mixture of MeOH/H₂O was added NaOH (0.48 g, 12.0 mmol), and the mixture was heated at 80°C for 16 h. The solvents were evaporated under reduced pressure, the residue was dissolved in THF (50 ml) and filtered. The filtrate was concentrated and the crude **6** was dissolved in THF (50 ml) followed by the addition of Et₃N (0.6 ml, 4.4 mmol) and Boc₂O (1.05 g, 4.8 mmol). The mixture was stirred for 16 h, concentrated to dryness and the residue was purified by silica gel chromatography

(CHCl₃/MeOH, 19:1) to give **7** (0.798 g, 65% yield): $[\alpha]_D = -20.0$ ($c = 1.88$, MeOH). IR (neat): $\nu = 3410$ cm⁻¹, 2927, 1696, 1393, 1253, 1171, 1114. – ¹H NMR: $\delta = 7.42$ – 7.22 (m, 5 H, Ph), 4.56, 4.52 (d, $J = 12.0$ Hz, 1 H each, OCH₂Ph), 4.45 (q, $J = 6.5$ Hz, 1 H, H-3 α), 3.95 (m, 1 H, H-2 α), 3.86, 3.76 (m, 1 H each, CH₂OCH₂Ph), 3.42 (t, $J = 7.0$ Hz, 2H, CH₂N), 2.04 (m, 1 H, CH_AH_B), 1.93 (dq, $J = 14.0, 3 \times 7.0$ Hz, 1 H, CH_AH_B), 1.45 [s, 9H, (CH₃)₃]. – ¹³C NMR: $\delta = 154.40$ (s, NCO), 137.35, 128.44, 127.80, 127.52 (s, 2 \times d, d, 2 \times d, Ph), 79.51 [s, OC(CH₃)₃], 73.48 (t, OCH₂Ph), 72.74, 72.23 (d each, CHO), 68.49 (t, CH₂OCH₂Ph), 58.70 (d, CHN), 44.39, 43.77 (t each, CH₂N), 32.89, 32.29 (t each, CH₂), 28.37 [q, (CH₃)₃]. Anal. Calcd. for C₁₇H₂₅NO₄ (307.4): C, 66.42; H, 8.20; N, 4.56. Found: C, 66.49; H, 7.89; N, 4.64.

(2*R*,3*R*)-1-Benzoyloxy-2-(tert-butoxycarbonyl)amino-3,5-pentandiol (8): A solution of oxazolidin-2-one **3** (1.92 g, 7.6 mmol) in EtOH (600 ml) was added to a satd. solution of Ba(OH)₂ (140 ml) and refluxed overnight. After that time most of the organic solvent was removed under reduced pressure and resulting aqueous solution was poured on the top of an ion-exchange column (Amberlist IR 120) and eluted with H₂O, followed by 30% NH₄OH. The fractions with the amino alcohol were concentrated under reduced pressure and the residue dissolved in THF (100 ml) followed by the addition of Boc₂O (2.18 g, 10.0 mmol) and Et₃N (1.17 ml, 8.4 mmol). The mixture was stirred overnight at room temp. concentrated under reduced pressure and purified by flash column chromatography (Et₂O/EtOAc 3:1) to give **8** (1.68 g, 68% yield): $[\alpha]_D = +27.8$ ($c = 0.13$, MeOH). IR (neat): $\nu = 3298$ cm⁻¹, 2973, 2929, 1686, 1533, 1367, 1169. – ¹H NMR: $\delta = 7.40$ – 7.25 (m, 5 H, Ph), 5.28 (brd, $J = 8.0$ Hz, 1 H, NH), 4.56 (brs, 2H, OCH₂Ph), 4.18 (m, 1 H, CHO), 3.93, 3.86 (m, 1 H each, CH₂O), 3.84, 3.77 (m, 1 H each, CH₂OH), 3.64 (m, 1 H, CHN), 2.80 (brs, 2H, 2 \times OH), 1.78, 1.66 (m, 1 H each, CH₂), 1.44 [s, 9H, (CH₃)₃]. – ¹³C NMR: $\delta = 156.39$ (s, NCO), 137.54, 128.44, 128.03, 127.75 (s, 2 \times d, 2 \times d, d, Ph), 80.07 [s, OC(CH₃)₃], 73.42 (t, OCH₂Ph), 71.67 (d, CHO), 67.14 (t, CH₂O), 61.24 (t, CH₂OH), 53.73 (d, CHN), 35.36 (t, CH₂), 28.35 [q, (CH₃)₃]. Anal. Calcd. for C₁₇H₂₇NO₅ (325.41): C, 62.75; H, 8.36; N, 4.31. Found: C, 62.80; H, 8.43; N, 4.37.

(2*R*,3*R*)-1-Benzoyloxy-2-(tert-butoxycarbonyl)amino-5-[(β -naphthyl)sulphonyloxy]-3-pentanol (9): To a cooled (-10°C) solution of the alcohol **8** (0.428 g, 1.31 mmol) in dry pyridine (2 ml) was added in one portion the β -naphthalen-2-sulphonyl chloride (0.400 g, 1.76 mmol) and the mixture was stirred at the same temperature for 3 h. Standard work-up gave the crude sulphonate, which was purified by silica gel column chromatography (*n*-hexane/EtOAc 1:1) to give **9** (0.445 g, 67%); $[\alpha]_D = +14.4$ ($c = 0.16$, MeOH). – IR (KBr): $\nu = 3360$ cm⁻¹, 2920, 1696, 1405, 1200, 1096, 1032. – ¹H NMR (CDCl₃/CD₃SOCD₃, 7:1): $\delta = 8.38$ (d, $J = 2.0$ Hz, 1 H, H-1' Naphth), 7.93 (dd, $J = 8.0, 2.0$ Hz, 1 H, H-3' Naphth), 7.88–7.80 (m, 2H, H-5', H-8' Naphth), 7.81 (d, $J = 8.0$ Hz, 1 H, H-4' Naphth), 7.50, 7.48 (dt, $J = 2 \times 7.0, 2.0$ Hz, 1 H each, H-6', H-7' Naphth), 7.35–7.25 (m, 5 H, Ph), 5.15 (brs, 1 H, NH), 4.55, 4.49 (d, $J = 11.0$ Hz, 1 H each, OCH₂Ph), 4.46 (m, 1 H, CHO), 3.92 (dd, $J = 11.0, 8.0$ Hz, 1 H, CH_AH_BO), 3.87 (dd, $J = 11.0, 6.0$ Hz, 1 H, CH_AH_BO); 3.66 (ddd, $J = 8.0$ Hz, 5.0 Hz and 4.0 Hz, 1 H, CHN), 3.59 (m, 1 H, CHCH_DOSO₂), 3.40 (ddd, $J = 12.0, 9.0, 4.0$ Hz, 1 H, CHCH_DOSO₂), 2.18, 2.11 (m, 1 H each, CH₂), 1.44 [s, 9H, (CH₃)₃]. – ¹³C NMR (CDCl₃/CD₃SOCD₃, 7:1): $\delta = 154.84$ (s, NCO), 142.60 (s, C-2' Naphth), 137.47 (s, C-1" Ph), 133.66, 132.41 (s each, C-1', C-4' Naphth), 128.75, 127.60 (d each, C-5', C-8' Naphth), 128.33 (2 \times d, C-3", C-5" Naphth), 128.07 (3 \times d, C-2", C-6", Ph; C-8' Naphth), 127.79 (d, C-4", Ph), 126.98, 126.46 (d each, C-6', C-7' Naphth), 125.36 (d, C-1' Naphth), 123.39 (d, C-3' Naphth), 80.08 [s, OC(CH₃)₃], 73.38 (t, OCH₂Ph), 70.12 (d, CHO), 66.45 (t, CH₂O), 53.71 (d, CHN), 43.43 (t, CH₂OS), 33.46 (t, CH₂), 28.40 [q, (CH₃)₃]. Anal. Calcd. for C₂₇H₃₃NO₇S (515.64): C, 62.90; H, 6.45; N, 2.72. Found: C, 62.69; H, 6.39; N, 2.69.

(2R,3R)-1-(tert-Butoxycarbonyl)-2-(benzyloxymethyl)-3-hydroxypyrrolidine (7): To a cooled (0°C) suspension of NaH (60%, 44 mg, 1.1 mmol, washed three times with petroleum ether) in THF (20 ml) was added dropwise the sulphonate **9** (0.515 g, 1.0 mmol) in THF (5 ml). The mixture was stirred at the same temperature for 3 h, then H₂O was added cautiously. Most of the organic solvent was removed under reduced pressure and the residue was extracted with CHCl₃ (3 × 50 ml). Standard work-up gave **7** (0.264 g, 86% yield).

(2R,3R)-1-(tert-Butoxycarbonyl)-2-(benzyloxymethyl)-3-[(tert-butyldimethylsilyloxy]-pyrrolidine (10): A mixture of (2*S*,3*R*)-**7** (0.614 g, 2.0 mmol), imidazole (0.340 g, 5.0 mmol), DMAP (49 mg, 0.40 mmol), and TBDMSCl (0.452 g, 3.0 mmol) in DMF (10 ml) was stirred at room temperature for 24 h. The solvent was evaporated under reduced pressure, the residue was dissolved in EtOAc (25 ml) and filtered. The organic solvent was removed and the residue was purified by flash chromatography over silica gel eluting with *n*-hexane/EtOAc (6:4) to give **10** (0.639 g, 76% yield) as a colorless oil: $[\alpha]_D = -20.0$ ($c = 0.28$, CHCl₃). – IR (neat): $\nu = 2934$ cm⁻¹, 2861, 1696, 1393, 1253, 1122. – ¹H NMR: $\delta = 7.35$ – 7.20 (m, 5 H, Ph), 4.53, 4.47 (d, $J = 12.0$ Hz, 1 H each, OCH₂Ph), 4.37 (dt, $J = 8.0, 2 \times 6.0$ Hz, 1 H, CHO), 3.82–3.68 (m, 2H, CH₂O), 3.72 (m, 1 H, CHN), 3.41 (ddd, $J = 11.0, 8.0, 5.0$ Hz, 1 H, CHAHBN), 3.33 (dt, $J = 11.0, 2 \times 8.0$ Hz, 1 H, CHAHBN), 1.99 (dq, $J = 12.0, 3 \times 8.0$ Hz, 1 H, CHC_HD), 1.90 (dddd, $J = 11.0, 8.0, 6.0, 5.0$ Hz, 1 H, CHC_HD), 1.43 [s, 9H, (CH₃)₃], 0.89 [s, 9H, SiC(CH₃)₃], 0.07, 0.06 (s, 3H each, Si(CH₃)₂). — ¹³C NMR: 154.64 (s, NCO), 138.92, 128.03, 127.23, 127.05 [s, 2 × d, 2 × d, d, Ph], 79.11 [s, OC(CH₃)], 73.10 (t, OCH₂Ph), 71.72 (d, CHO), 67.49 (t, CH₂O), 59.76 (d, CHN), 43.77 (t, CH₂N), 32.53 (t, CH₂), 28.40 [q, (CH₃)₃], 25.69 [q, SiC(CH₃)₃], 17.96 (s, SiC), –4.90, –5.14 [q each, Si(CH₃)₂]. Anal. Calcd. for C₂₃H₃₉SiNO₄ (421.7): C, 65.51; H, 3.32; N, 3.39. Found: C, 65.62; H, 3.37; N, 3.29.

(2R,3R)-1-(tert-Butoxycarbonyl)-2-(hydroxymethyl)-3-[(tert-butyldimethylsilyloxy]-pyrrolidine (11): A solution of compound **10** (0.696 g, 1.65 mmol) was dissolved in abs. EtOH (25 ml) and hydrogenated over 10% Pd/C at 1 atm for 16 h. After that time the mixture was filtered over Celite, concentrated and purified by silica gel chromatography eluting with CHCl₃/MeOH (20:1) to give **11** (0.453 g, 83% yield) as an oil: $[\alpha]_D = -33.1$ ($c = 1.2$, CHCl₃) (Lit.²⁰ $[\alpha]_D = -32.5$ ($c = 1.06$, CHCl₃)). – ¹H and ¹³C NMR as reported.²⁰

(2*S*,3*R*)-3-Hydroxyproline (1): A solution of **11** (0.422 g, 1.0 mmol), TEMPO (1.5 mg, 0.01 mmol) and NaBr (0.103 g, 1.0 mmol) in toluene (0.8 ml), EtOAc (0.8 ml) and H₂O (0.05 ml) was cooled at 0°C. To the rapid mechanically stirring mixture, an aqueous solution of NaOCl (0.35 M, 3.14 ml, 1.1 mmol) containing NaHCO₃ (269 mg, 32.0 mmol) was added dropwise over a period of 30 min, and stirred for an additional 10 min. The aqueous layer was separated and washed with 10 % aqueous KHSO₄ (10 ml) containing KI (0.06 g). The organic layer was then washed with aqueous Na₂S₂O₄ (0.2 M, 20 ml), brine, dried (Na₂SO₄), and concentrated to dryness. The resulting aldehyde was dissolved in *tert*-butyl alcohol (2.3 ml) and treated with 5% NaH₂PO₄ (1.53 ml) and KMnO₄ (1 M, 2.2 ml) solutions. After 30 min the solution was diluted with Et₂O (10 ml) and cooled at 0°C. A saturated solution of Na₂SO₃ (1 ml) was added dropwise and acidified to pH 3 with dilute HCl. The aqueous layer was extracted with EtOAc (3 × 50 ml). The combined organic layers were dried (Na₂SO₄), filtered, and concentrated to dryness.

To an ice-cooled solution of the above acid in MeOH (2 ml) a satd. solution of HCl in MeOH (3 ml) was added. The mixture was stirred at room temp. for 2 h and then concentrated under reduced pressure. The residue was dissolved in H₂O (2 ml) and purified on an ion-exchange column (DOWEX 50 × 8, 200–400 mesh) using 1.5 M NH₃ as eluent to give 82 mg (63%) of **1** as white crystals; m.p. 222–228°C (dec.), $[\alpha]_D = -100.4$ ($c = 1.0$, H₂O); (Lit.^{14c} m.p. 220–230°C, $[\alpha]_D = -101$ ($c = 1.0$, H₂O)).

(2*R*,3*S*)-1-(*tert*-Butoxycarbonyl)-2-(benzyloxymethyl)-3-[(4-nitrobenzoyl)oxy]pyrrolidine (12): To a stirred solution of the alcohol **7** (0.460 g, 1.5 mmol), triphenylphosphine (1.96 g, 7.5 mmol) and *p*-nitro benzoic acid (1.1 g, 5.72 mmol) in benzene (30 ml) at room temperature, diethylazodicarboxylate (1.17 ml, 7.5 mmol) was added dropwise. The slightly orange solution was stirred at room temperature for 6 h, whereupon the volatile components were removed under reduced pressure and the residue purified by silica gel chromatography eluting with *n*-hexane/EtOAc (6:4) to give **12** (0.616 g, 90% yield); $[\alpha]_{\text{D}} = +13.52$ ($c = 0.15$, CHCl₃). – IR (neat): $\nu = 2937 \text{ cm}^{-1}$, 1720, 1657, 1412, 1189, 1106. – ¹H NMR: $\delta = 8.29, 8.19$ (d, $J = 9.0$ Hz, 2H each, *p*-NO₂Ph), 7.40–7.25 (m, 5 H, Ph), 5.57 (dd, $J = 12.0, 5.0$ Hz, 1 H, CHO), 4.59, 4.54 (d, $J = 12.0$ Hz, 1 H each, OCH₂Ph), 4.18 (m, 1 H, CHN), 3.76, 3.68 (m, 1 H each, CH₂O), 3.60, 3.52 (m, 1 H, CH₂N), 2.41 (m, 1 H, CH_AH_B), 2.12 (brdt, $J = 14.0, 2 \times 5.0$ Hz, 1 H, CH_AH_B), 1.49 [s, 9H, (CH₃)₃]. – ¹³C NMR: $\delta = 164.10$ (s, OCO), 154.21 (s, NCO), 150.60, 135.51, 130.77, 123.54 (s, s, 2 × d, 2 × d, *p*-NO₂Ph), 137.83, 128.43, 127.76, 127.46 (s, 2 × d, d, 2 × d, Ph), 80.07, 79.87 [s each, OC(CH₃)₃], 79.00, 78.35 (d each, CHO), 73.37 (t, OCH₂Ph), 69.27, 68.23 (t each, CH₂O), 63.07 (d, CHN), 45.07, 44.75 (t each, CH₂N), 30.33, 29.36 (t, CH₂), 28.44 [q, (CH₃)₃]. Anal. Calcd. for C₂₄H₂₈N₂O₄ (456.5): C, 63.15; H, 6.18; N, 6.14. Found: C, 63.02; H, 6.07; N, 6.24

(2*R*,3*S*)-1-(*tert*-Butoxycarbonyl)-2-(hydroxymethyl)-3-hydroxypyrrolidine (13): To a solution of NaOH (0.150 g, 3.75 mmol) in MeOH (30 ml) was added the *p*-nitro benzoate **12** (0.330 g, 0.72 mmol). The mixture reaction was stirred at room temp. overnight, evaporated to dryness, and H₂O (10 ml) was added. The aqueous mixture was extracted with EtOAc (3 × 50 ml), and the combined organic extracts were washed with brine, dried (Na₂SO₄) and the solvent removed. The residue was purified by silica gel column chromatography (CHCl₃/MeOH, 20:1) to give **13** (0.199 g, 90% yield) as a white solid. M.p. 67–8°C, $[\alpha]_{\text{D}} = -33.4$ ($c = 0.18$, MeOH). – IR(KBr): $\nu = 3394 \text{ cm}^{-1}$, 2918, 1655, 1417, 1171, 1106. – ¹H NMR: $\delta = 7.37$ –7.24 (m, 5 H, Ph), 4.54, 4.50 (d, $J = 12.0$ Hz, 1 H each, OCH₂Ph), 4.39 (ddd, $J = 5.0, 3.0, 1.5$ Hz, 1 H, H-3β), 3.78, 3.37 (m, 1 H each, CH₂O), 3.70, 3.25 (m, 1 H each, CH₂N), 3.56 (m, 1 H, CHCH₂O), 2.05 (ddt, $J = 13.5, 2 \times 9.0, 5.0$ Hz, 1 H, CHCH₂CH₂), 1.82 (ddt, $J = 13.5, 7.5, 2 \times 3.0$ Hz, 1 H, CHCH₂CH₂), 1.44 [s, 9 H, (CH₃)₃]. ¹³C NMR: $\delta = 154.69$ (s, NCO), 138.30, 128.44, 127.70, 127.56 (s, 2 × d, d, 2 × d, Ph), 79.55 (s, OC), 74.83, 74.05 (d each, CHO), 73.49 (t, OCH₂Ph), 70.24 (t, CH₂O), 65.34 (d, CHN), 44.65, 44.55 (t each, CH₂N), 31.96 (t, CH₂), 28.53 [q, (CH₃)₃]. Anal. Calcd. for C₁₇H₂₅NO₄ (307.4): C, 66.42; H, 8.20; N, 4.56. Found: C, 66.31; H, 8.10; N, 4.64.

(2*R*,3*S*)-1-(*tert*-Butoxycarbonyl)-2-hydroxymethyl-3-hydroxypyrrolidine (14): Compound **13** (0.250 g, 0.82 mmol) was dissolved in absolute EtOH (20 ml) and hydrogenated over 10% Pd/C at 1 atm for 18 h. After that time the solution was filtered over Celite and concentrated under reduced pressure to give **14** (0.173 g, 98% yield) as a colorless oil, which was used without further purification. $[\alpha]_{\text{D}} = -21.5$ ($c = 0.66$, MeOH). – IR (neat): $\nu = 3394 \text{ cm}^{-1}$, 2976, 1671, 1417, 1171, 1122. – ¹H NMR: $\delta = 4.21$ (m, 1 H, CHO), 3.76 (m, 1 H, CHN), 3.68, 3.52 (m, 1 H each, CH₂O), 3.60 (m, 1 H, CH_AH_BN), 3.38 (ddd, $J = 11.0, 8.5, 4.0$ Hz, 1 H, CH_AH_BN), 2.03 (m, 1 H, CH₂CH_D), 1.86 (ddt, $J = 14.0, 7.0, 2 \times 3.5$ Hz, 1 H, CH₂CH_D), 1.46 [s, 9 H, (CH₃)₃]. – ¹³C NMR: $\delta = 156.40$ (s, NCO), 80.34 (s, OC), 72.72 (d, CHO), 67.94 (d, CHN), 64.10 (t, CH₂O), 44.96 (t, CH₂N), 31.84 (t, CH₂), 28.40 [q, (CH₃)₃]. Anal. Calcd. for C₁₀H₁₉NO₄ (217.3): C, 55.28; H, 8.82; N, 6.45. Found: C, 55.31; H, 8.71; N, 6.30.

(2*R*,3*S*)-2-Hydroxymethyl-3-hydroxypyrrolidine Hydrochloride (2): Compound **14** (0.150 g, 0.7 mmol) was dissolved in 3 M HCl-EtOAc (1ml). After 30 min the solution was removed under reduced pressure and the oil was triturated with Et₂O to give **2** (97 mg, 91%): m.p. 110–112 °C, $[\alpha]_{\text{D}} = +43.1$ ($c = 0.5$, H₂O) (Lit.¹³: m.p. 108–112 °C, $[\alpha]_{\text{D}} = +46.5$ ($c = 1$, H₂O)). ¹H and ¹³C NMR as reported.¹³

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